## **Oximidines I and II: Novel Antitumor Macrolides from** Pseudomonas sp.

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Our screening for cell-cycle inhibitors in transformed cells resulted in the isolation of novel active compounds, oximidines I (1) and II (2), from *Pseudomonas* sp. Q52002. The molecular formulas of **1** and **2** were established as  $C_{23}H_{24}N_2O_7$  and  $C_{23}H_{24}N_2O_6$ , respectively, by high-resolution FABMS. The structures of the oximidines were elucidated by NMR spectral analysis, including a variety of two-dimensional techniques. The absolute stereochemistry of 1 was determined by using the modified Mosher method. The oximidines are novel 12-membered macrolides containing an O-methyloxime moiety. Oximidines I and II selectively inhibited the growth of rat 3Y1 cells transformed with E1A, ras, or src oncogenes.

Many types of oncogenes are implicated in cell-cycle control.<sup>1–3</sup> Thus, cell-cycle inhibitors in oncogene-transformed cells may act as selective antitumor drugs. Our screening for such antitumor substances resulted in the isolation of novel active metabolites, oximidines I (1) and II (2), from *Pseudomonas* sp. Q52002. We report herein the structures and biological activities of 1 and 2.

The molecular formulas of 1 and 2 were established as  $C_{23}H_{24}N_2O_7$  and  $C_{23}H_{24}N_2O_6$ , respectively, by highresolution FABMS. <sup>13</sup>C and <sup>1</sup>H NMR data for 1 and 2 are summarized in Table 1.

A series of NMR experiments including COSY and HMQC were used to construct three partial structures of 1, depicted in Figure 2 by bold lines. The partial structure A was extended to be a 2,3-disubstituted phenol based on an HMBC experiment, which revealed longrange couplings from H-4 to C-2, H-5 to C-3 and C-7, H-6 to C-2, and a phenolic hydroxyl proton to C-2, C-3, and C-4. In the partial structure B, the presence of an epoxide between C-12 and C-13 was required by their characteristic <sup>13</sup>C and <sup>1</sup>H chemical shifts (Table 1). The geometrical configurations were determined to be 8*E*, 10*Z*, 17*Z*, and 20Z from their relevant proton coupling constants ( $J_{8-9}$ = 16.5 Hz,  $J_{10-11}$  = 11.5 Hz,  $J_{17-18}$  = 9.0 Hz, and  $J_{20-21}$ = 11.5 Hz).

Long-range correlations, from H-8 to C-2, C-6, and C-7 and from H-9 to C-7, established the connection of the partial structures A and B (Figure 2). Two aromatic protons (H-4 and H-6) displayed four-bond correlations to a carbonyl carbon (C-1), indicating its substitution at C-2. A long-range coupling from H-15 to C-1 identified an ester linkage between C-1 and C-15 which leads to a 12-membered lactone ring. The partial structures B and C were joined via an amide carbonyl (C-19) as determined



Figure 1. Structures of oximidines I (1) and II (2).

Table 1. <sup>13</sup>C (125 MHz) and <sup>1</sup>H (500 MHz) NMR Data for Oximidines I (1) and II (2) in CDCl<sub>3</sub>

	1		2	
no.	$\delta_{\rm C}$	$\delta_{ m H}$ (multiplicity, Hz)	$\delta_{\rm C}$	$\delta_{ m H}$ (multiplicity, Hz)
1	169.8		171.0	
2	114.2		109.9	
3	159.5		162.7	
4	116.5	6.88 (d, 8.0)	117.5	6.89 (d, 8.0)
5	133.4	7.29 (dd, 8.0, 7.5)	134.1	7.31 (dd, 8.0, 7.5)
6	120.1	6.70 (d, 7.5)	121.7	6.73 (d, 7.5)
7	140.8		141.0	
8	134.8	6.96 (d, 16.5)	130.9	6.84 (d, 16.5)
9	128.0	5.96 (dd, 16.5, 7.0)	132.9	6.70 (dd, 16.5, 9.5)
10	129.1	6.10 (ddd, 11.5, 7.0, 1.5)	130.1	6.39 (dd, 10.5, 9.5)
11	124.4	5.69 (d, 11.5)	130.7	6.28 (d, 10.5)
12	57.6	3.90 (m)	127.1	6.08 (d, 12.5)
13	57.7	3.49 (dd, 8.5, 3.5)	133.3	5.71 (dd, 12.5, 3.0)
14	67.5	3.69 (dd, 8.5, 5.5)	72.9	4.32 (br d, 11.0)
15	74.9	5.05 (dt, 5.5, 5.0)	77.6	5.27 (dd, 10.5, 4.5)
16	25.4	2.85 (m)	27.8	2.59 (m)
		2.47 (m)		2.26 (m)
17	108.5	5.11 (dt, 8.5, 8.0)	104.5	4.80 (dt, 9.5, 8.0)
18	123.4	6.81 (dd, 11.0, 8.5)	126.1	6.91 (dd, 10.0, 9.5)
19	162.4		162.1	
20	124.7	5.92 (d, 11.5)	124.5	5.90 (d, 11.5)
21	135.2	6.44 (dd, 11.5, 10.5)	135.5	6.49 (dd, 11.5, 10.5)
22	147.7	8.97 (d, 10.5)	147.7	9.01 (d, 10.5)
OMe	62.3	3.87 (s)	62.3	3.92 (s)
3-OH		9.77 (s)		10.73 (s)
14-OH		3.37 (br)		3.17 (d, 11.0)
18-NH		8.65 (d, 11.0)		8.36 (d, 10.0)

from its long-range couplings to H-18, 18-NH, H-20, and H-21 (Figure 2).

The terminal sp<sup>2</sup> methine (C-22) in the partial structure C needed to be connected to the remaining nitrogen

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**Figure 2.** Connectivities among the partial structures of **1** by HMBC. Bold lines show proton spin systems, arrows indicate  ${}^{1}H{-}{}^{13}C$  long-range correlations, and a dashed line indicates an NOE correlation.



**Figure 3.** Absolute stereochemistry of **1**.  $\Delta \delta$  ( $\delta_{S-MTPA} - \delta_{R-MTPA}$ ) values are expressed in ppm. Dashed lines indicate NOEs.

atom in an imine-type bond. Four-bond  ${}^{1}H^{-13}C$  correlations from a methoxyl group to the imine carbon confirmed the presence of an *O*-methyloxime moiety. A NOESY experiment revealed a weak NOE between H-22 and the methoxyl group, indicating *E* geometry for the oxime double bond. The planar structure of **1** thus obtained is shown in Figure 2.

A cis configuration for the epoxide was established by a significant NOE and a 3.5-Hz coupling between H-12 and H-13. In addition, NOEs between H-13 and H<sub>2</sub>-16 indicated that H-12, H-13, and C-16 reside in the same orientation on the macrolide ring. A proton on C-14 displayed an NOE to H-11 but not to H-12 and was required to be in the orientation opposite that of H-12, thereby establishing the relative stereochemistry of **1** as shown in Figure 3.

The absolute stereochemistry of oximidine I (1) was elucidated by the modified Mosher method.<sup>4</sup> Both the (R)- and (S)-2-methoxy-2-(trifluoromethyl)phenyl acetate (MTPA) esters of 1 were prepared and subjected to <sup>1</sup>H NMR analysis. Only the two epoxide protons (H-12 and H-13) appeared downfield in the (R)-derivative spectrum relative to the (S)-derivative one, while most of the other protons including H-15, H<sub>2</sub>-16, H-17, and H-18 exhibited upfield shifts in the (R)-derivative spectrum. These data allowed the absolute configuration at C-14 to be assigned as R and completed the total structure of 1 as shown in Figure 1.

The molecular formula of **2**,  $C_{23}H_{24}N_2O_6$ , showed a loss of one oxygen in comparison with that of **1**. In the <sup>1</sup>H NMR spectrum of **2**, H-12 and H-13 appeared downfield (H-12,  $\delta$  3.90  $\rightarrow$  6.80; H-13,  $\delta$  3.49  $\rightarrow$  5.71), and the corresponding carbon signals were observed at  $\delta$  127.1 and  $\delta$  133.3, respectively, indicating that the expoxide



**Figure 4.** Structures of salicylihalamide A (**3**) and CJ-12,-950 (**4**).

Table 2.IC50 Values (ng/mL) of 1 and 2 against Normal<br/>and Transformed Rat 3Y1 Fibroblasts

cell line	oncogene	1	2
3Y1 SV-3Y1 E1A-3Y1 Ad12-3Y1 HR-3Y1 SR-3Y1	SV40 large T E1A E1A, E1B v-H- <i>ras</i> v- <i>src</i>	510 740 30 62 16 27	270 350 17 36 9.0 14

group in **1** is replaced by an olefinic double bond in **2**. The geometrical configuration of the new double bond was identified as Z on the basis of a significant NOE between H-12 and H-13, although their coupling constant exhibited a value (12.5 Hz) intermediate between those for E and Z configurations. The remaining spectral features were essentially identical with those of **1**. The structure of **2** (Figure 1), including the relative stereo-chemistry, was confirmed by COSY, HMQC, HMBC, and NOESY experiments (data not shown).

Recently, a series of 12-membered benzolactones have been isolated from natural sources, e.g., salicylihalamide A (**3**) and its geometrical isomer B from the marine sponge *Haliclona* sp.<sup>5</sup> and CJ-12,950 (**4**) and its geometrical isomer CJ-13,357 from the fungus *Mortierella verticillata* (Figure 4).<sup>6</sup> Oximidines I (**1**) and II (**2**) represent two new members of this family and are the first examples produced by a bacterium.

The antitumor effects of **1** and **2** were investigated using normal and transformed 3Y1 rat fibroblasts (Table 2).<sup>7–9</sup> Oximidines I and II inhibited the growth of 3Y1 cells transformed with the adenovirus E1A, v-H-*ras*, and v-*src* oncogenes at 15- to 30-fold lower concentrations than that of the parent 3Y1 cells, whereas SV40transformed cells showed almost the same sensitivity as normal cells. Flow cytometric analysis revealed that **1** arrested the cell cycle of *ras*- and *src*-transformed 3Y1 cells at the G1 phase (data not shown). Further biological studies on the oximidines are in progress.

## **Experimental Section**

**Production and Isolation.** The producing organism was inoculated into 500-mL Erlenmeyer flasks containing 100 mL of a production medium consisting of soluble starch 2%, glycerol 0.5%, soybean meal 1.5%, corn steep liquor 0.5%, NaCl

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0.3%, and CaCO<sub>3</sub> 0.3% (pH 7.0, before sterilization). The fermentation was carried out at 28 °C for 72 h on a rotary shaker at 200 rpm.

The culture supernatant (2 L) and the mycelial acetone extract were combined and extracted with EtOAc. The extract was chromatographed on a silica-gel column with CHCl<sub>3</sub>–MeOH (80:1) and then subjected to HPLC using a YMC-Pack D-ODS-7 column. Development of the column with 60% MeOH gave active fractions I and II. The fraction I was further purified by HPLC using a PEGASIL ODS column with 55% MeOH. The active fraction was evaporated to dryness, to yield a colorless powder of **1** (108 mg). The fraction II was subjected to silica-gel TLC with hexane–EtOAc (1:1). The active fraction was purified by Sephadex LH-20 column chromatography with CHCl<sub>3</sub>-MeOH (1:1), following which it was concentrated to dryness to give a colorless power of **2** (15 mg).

**Oximidine I (1):** mp 104–106 °C;  $[\alpha]^{22}_{D}$  –286° (*c* 1.0, MeOH); UV  $\lambda_{max}$  272 nm ( $\epsilon$  30500) in MeOH; 203 nm ( $\epsilon$  112000) and 270 nm (30500) in 0.01 M NaOH–MeOH; IR (KBr)  $\nu_{max}$  3350, 1710, and 1650 cm<sup>-1</sup>; FABMS *m*/*z* 441.1673 MH<sup>+</sup> (calcd for C<sub>23</sub>H<sub>25</sub>N<sub>2</sub>O<sub>7</sub>, 441.1662).

**Oximidine II (2):** mp 113–115 °C;  $[\alpha]^{26}_D$  –141° (*c* 0.75, MeOH); UV  $\lambda_{max}$  282 nm ( $\epsilon$  30300) in MeOH; 203 nm ( $\epsilon$  138000) and 279 nm (28200) in 0.01 M NaOH–MeOH; IR (KBr)  $\nu_{max}$  3400, 1700, and 1650 cm<sup>-1</sup>; FABMS *m*/*z* 425.1702 MH<sup>+</sup> (calcd for C<sub>23</sub>H<sub>25</sub>N<sub>2</sub>O<sub>6</sub>, 425.1713).

**Preparation of the MTPA Esters of 1.** To a solution of **1** (12 mg) in CHCl<sub>3</sub> (500 μL) were added 500 μL of pyridine, 0.1 mg of 4-(dimethylamino)pyridine, and 56 mg of (S)-(+)-α-methoxy-α-(trifluoromethyl)phenylacetyl chloride. The reaction mixture was allowed to stand at room temperature for 30 min. After 10 mL, each, of H<sub>2</sub>O and CHCl<sub>3</sub> were added, the CHCl<sub>3</sub> layer was washed with aqueous HCl and aqueous NaOH, dried over Na<sub>2</sub>SO<sub>4</sub>, and evaporated to dryness. The dried material was purified by HPLC using a YMC-Pack D-ODS-7 column with 60% MeOH. The major UV peak was collected and evaporated to dryness to give a colorless powder of the (*R*)-MTPA ester of **1** (3.2 mg). Similarly, the (*S*)-MTPA ester of **1** (4.3 mg) was obtained using (*R*)-(-)-α-methoxy-α-(trifluoromethyl)phenylacetyl chloride.

(*R*)-**MTPA ester of 1:** <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  9.81 (s, OH-3), 8.99 (d, J = 10.0 Hz, H-22), 7.88 (d, J = 11.0 Hz, 18-NH), 7.54

(d, J = 7.5 Hz, MTPA), 7.44 (t, J = 7.5 Hz, MTPA), 7.40 (dd, J = 8.0, 7.5 Hz, H-5), 7.40 (t, J = 7.5 Hz, MPTA), 7.07 (d, J = 16.0 Hz, H-8), 6.94 (d, J = 8.0 Hz, H-4), 6.75 (d, J = 7.5 Hz, H-6), 6.50 (dd, J = 11.0, 10.0 Hz, H-21), 6.38 (dd, J = 11.0, 8.5 Hz, H-18), 6.20 (ddd, J = 11.5, 6.0, 1.5 Hz, H-10), 5.86 (dd, J = 16.0, 6.0 Hz, H-9), 5.84 (d, J = 11.0 Hz, H-20), 5.81 (d, J = 11.5 Hz, H-11), 5.17 (dd, J = 9.0, 6.0 Hz, H-14), 4.90 (m, H-15), 4.41 (dt, J = 8.5, 8.0 Hz, H-17), 3.96 (m, H-12), 3.93 (s, OMe), 3.70 (s, MTPA-OMe), 3.62 (dd, J = 9.0, 3.5 Hz, H-13), 2.56 (m, H-16<sub>a</sub>), 2.40 (m, H-16<sub>b</sub>).

(S)-MTPA ester of 1: <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  9.86 (s, OH-3), 9.01 (d, J = 10.5 Hz, H-22), 8.20 (d, J = 11.0 Hz, 18-NH), 7.55 (d, J = 8.0 Hz, MTPA), 7.43 (t, J = 8.0 Hz, MTPA), 7.40 (dd, J = 8.0, 7.5 Hz, H-5), 7.40 (t, J = 8.0 Hz, MTPA), 7.08 (d, J =16.5 Hz, H-8), 6.95 (d, J = 8.0 Hz, H-4), 6.81 (dd, J = 11.0, 8.5 Hz, H-18), 6.76 (d, J = 7.5 Hz, H-6), 6.52 (dd, J = 11.5, 10.5 Hz, H-21), 6.20 (ddd, J = 11.5, 6.0, 1.5 Hz, H-10), 5.91 (d, J =11.5 Hz, H-20), 5.86 (dd, J = 16.5, 6.0 Hz, H-9), 5.82 (d, J =11.5 Hz, H-11), 5.22 (dd, J = 9.0, 6.0 Hz, H-14), 5.05 (m, H-15), 4.80 (dt, J = 8.5, 8.0 Hz, H-17), 3.93 (s, OMe), 3.93 (m, H-12), 3.57 (dd, J = 9.0, 3.5 Hz, H-13), 3.48 (s, MTPA–OMe), 2.74 (m, H-16<sub>a</sub>), 2.64 (m, H-16<sub>b</sub>).

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**Supporting Information Available:** Spectral data (<sup>1</sup>H NMR, <sup>13</sup>C NMR, COSY, HMQC, HMBC, and NOESY) are available for oximidines I and II in CDCl<sub>3</sub> (12 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.

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